



F.E. (REV) (CBSGS), Sem-II

Sub: Applied Chemistry-II

(REVISED COURSE) Q.P. Code : 1049

(2 Hours)

[ Total Marks : 60

- N.B. : (1) Question No. 1 is Compulsory.  
(2) Attempt any three from remaining fix questions  
(3) All questions carry equal marks.  
(4) Figures to the right indicate full marks.  
(5) Atomic weights : H=1, C=12, S=32, N=14, O=16 ,Cl=35.5, Ba=137.3, Na=23, Mg=24.

1. Answer any five from the following :- 15
- (a) What are propellants ? State important characteristics of good propellant.
  - (b) Compare Galvanizing and Tinning.
  - (c) Give composition, properties and uses of Wood's Metal.
  - (d) Write a note on 'Green Reagent'.
  - (e) Define terms :-
    - (i) Composite material
    - (ii) Matrix phase
    - (iii) Dispersed phase.
  - (f) List three main constituents of Varnish & give functions of each.
  - (g) A coal sample was subjected to ultimate analysis.  
1.6 gm of coal on combustion in a Bomb calorimeter gave 0.47 gm of BaSO<sub>4</sub>  
Calculate % of sulphur in the coal sample.
2. (a) What is dry corrosion ? Explain with example how nature of oxidised product affect the rate of corrosion. 6
- (b) What is cracking ? Explain fixed bed catalytic cracking with diagram. 5
- (c) Calculate percentage atom economy for the following reaction w.r. to methyl iso-cyanate 4
- $$\text{CH}_3\text{NH}_2 + \text{COCl}_2 \rightarrow \text{CH}_3\text{-N=C=O} + 2\text{HCl}$$
- methyl iso cyanate.
3. (a) A gaseous fuel has the following composition by volume. 6  
CH<sub>4</sub> = 35%, C<sub>2</sub>H<sub>4</sub> = 5%, CO = 15%, H<sub>2</sub> = 40% N<sub>2</sub> = 1% water vapour = 4%  
Calculate volume & weight of air required for complete combustion of 1m<sup>3</sup> of fuel  
[mol.wt of air = 28.94]
- (b) Explain conventional & green synthesis of adipic acid. Mention the green chemistry principle involved. 5
- (c) How the rate of corrosion influenced by following factors. 4
  - (i) PH of medium
  - (ii) Over voltage.
4. (a) What is powder Metallurgy ? How are metal powders prepared using. 6
  - (i) Atomization
  - (ii) Chemical reduction
- (b) What is cathodic protection ? Explain Impressed current method of corrosion control. 5

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- (c) Write a note on 'Sandwich panel' type layered composite. 4
5. (a) What is Bio-diesel ? Explain the trans esterification reaction of Bio-diesel. Mention advantages of biodiesel as fuel. 5  
(b) What are alloys ? Explain any four purposes of making alloys with examples. 5  
(c) Discuss the physical factors influencing adhesive action. 4
6. (a) Write a note on differential aeration corrosion. 3  
(b) 2.5 gm of air dried coal sample was taken in a silica crucible, heated in electric oven at  $110^{\circ}\text{C}$  for 1hr the residue was weighed 2.41 gm. The residue was heated in Silica crucible covered with vented lid at a temperature of  $900^{\circ}\text{C}$  for exactly 7 minutes. After cooling the weight of residue was found to be 1.94 gm. The residue was then ignited to a constant weight of 1.48 gm. Report the results of above analysis. 5  
(c) Explain the effects of following elements on alloying. 5  
(i) Nickel  
(ii) Chromium  
(iii) Cobalt  
(iv) Molybdenum  
(v) Carbon.



